

**(Ferrozine Method)**

**CLINICAL SIGNIFICANCE:**

Measurement of Unsaturated Iron-binding Capacity (UIBC) are used to assist in the diagnosis and treatment of **anaemia**.

**PRINCIPLE:**

Excess ferrous ions bind specifically with available iron-binding sites of transferrin and saturating the molecules with iron in alkaline medium.

Ferrozine reacts with the remaining unbound iron to form a strongly purple coloured complex which is measured at 560 nm. The difference between the known excess amount of iron added and the remaining unbound iron is equivalent to unsaturated iron-binding capacity (UIBC). Total iron binding capacity (TIBC) is calculated as serum iron plus UIBC.

**REAGENT COMPOSITION:**

Active Ingredients	Concentration
<b>Reagent-1</b>	
* Buffer	100 mmol/L
* Detergent	5 mmol/L
pH 4.5 ± 0.1 at 25° C	
<b>Reagent-2</b>	
* Ferrozine	4 mmol/L
* Hydroxylamine Hydrochloride	150 mmol/L
<b>Reagent-3</b>	
* Enhancer	10 mmol/L

**UIBC Standard (80 µmol/L)**

Also contains non-reactive fillers and Stabilizers.

**PRESENTATION:**

	No. of Bottles
Store all reagents at 2-8°C	20 Test
1 UIBC (Buffer Reagent)	2
2 UIBC (Colour Reagent)	1
3 UIBC (Enhancer)	2
UIBC standard (80 µmol/L)	1

**PREPARATION OF WORKING REAGENT -1:**

Carefully transfer the content of one vial of 3 UIBC to one bottle of 1 UIBC. Mix to dissolve. Wait for 15 minutes before use.

**REAGENT STORAGE AND STABILITY:**

When stored, refrigerated at 2-8°C and protected from light, the reagents are stable until the expiry date stated on the bottle and kit box labels. The working reagent is stable for 2 months at 2-8°C.

**SPECIMEN COLLECTION AND HANDLING:**

Fresh clear serum or plasma with no hemolysis should be used. Specimens are stable for one day at room temperature or one week at 2-8°C.

**PRECAUTION:**

It is essential that all the glasswares used for assay should be Iron-free. Glasswares should be soaked in 0.1N HNO<sub>3</sub> or HCl & rinsed thoroughly with iron-free deionized water.

**REACTION PARAMETERS :**

Monochromatic	
Type of reaction	: End point (Increase)
Wavelength	: 560 nm
Temperature	: 37°C
Incubation	: 10 min at 37°C
Std. Concentration	: 80 µmol/L (446 µg/dL)
Std./Sample Volume	: 200 µl (0.200 ml)
Reagent 1	: 1.0 ml
Reagent 2	: 0.050 ml (50 µl)
Light path	: 1.0 cm
Zero setting with	: Reagent blank

**Bichromatic**

Other parameters as above

Wavelength	: 560 nm and 630 nm
Sample Blank	: No
Zero setting with	: Distilled water
Set the instrument using above system parameters	

**PROCEDURE:**

**A) Monochromatic Method :**

Pipette into test tube	Blank	Std.	Test	
			Sample Blank (A1)	Sample Test (A2)
Rgt.-1 (ml)	1.0	1.0	1.0	1.0
Sample (ml)	-	-	0.2	0.2
Standard (ml)	-	0.2	0.2	0.2
Dist. Water (ml)	0.4	0.2	-	-
Rgt.-2 (ml)	0.05	0.05	-	0.05

Mix and allow to stand for 10 minutes .at 37°C. Read absorbance of test (A<sub>1</sub>& A<sub>2</sub>) and standard against reagent blank at 560 nm.

**B) Bichromatic Method:**

Pipette into Test tube	Std.	Test
Reagent -1 (ml)	1.0	1.0
Sample (ml)	-	0.2
Standard (ml)	0.2	0.2
Dist. Water (ml)	0.2	-
Reagent-2 (ml)	0.05	0.05

Mix and allow to stand for 10 minutes .at 37°C. Read absorbance of test and standard against distilled water at 560 nm & 630 nm.

**TEST RESULTS:**

$$\text{Excess Iron } (\mu\text{mol/L}) = \frac{\text{Abs or } \Delta \text{ Absorbance of test} \times 80 (\mu\text{mol/L})}{\text{Absorbance of Std.}}$$

Where

$$\Delta \text{ Absorbance} = (A_2 - A_1) \\ 80 \mu\text{mol/L} = \text{Concentration of Standard}$$

$$\text{UIBC } (\mu\text{mol/L}) = 80 - \text{excess Iron } (\mu\text{mol/L}).$$

$$\text{TIBC } (\mu\text{mol/L}) = \text{Serum Iron } (\mu\text{mol/L}) + \text{UIBC } (\mu\text{mol/L}).$$

$$\text{To convert } (\mu\text{g/dl}) = \mu\text{mol/L} \times 5.585$$

**EXPECTED VALUES:**

$$\text{UIBC: } 28.6 - 64.5 \mu\text{mol/L (160-360 } \mu\text{g/dl)}$$

$$\text{TIBC: } 44.7 - 71.6 \mu\text{mol/L (250-400 } \mu\text{g/dl)}$$

**LIMITATIONS:**

- 1) Hemolysis causes falsely elevated results.
- 2) Iron medications (oral, intravenous or intravascular) affect serum levels.

**LINEARITY:**

This procedure is linear upto 89 µmol/L (500 µg/dl). For sample values higher than 89 µmol/L (500 µg/dl), dilute the sample suitably with 0.9% saline and repeat the assay. Apply dilution factor to obtain test results.

**REFERENCES:**

1. Tietz NW "Text book of clinical chemistry 2nd Edition" Tetz NW (Ed) WB Saunders company Philadelphia 1994; 2059.
2. CaO G.and Prior R.L. Chemistry Anthocyanins and iron metabolism in human serum 1999b; 574-76.
3. National committee for Clinical Laboratory Standards. User evaluation of precision performance of clinical chemistry Devices. NCCLS, 1984 NCCLS Publication EP5-T.



